

Solid Oxide Fuel Cell Structure and Application Research

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Abstract:

In the context of accelerating the global promotion of carbon peak and carbon neutrality, solid oxide fuel cells (SOFC) are gradually becoming an important direction of clean energy technology thanks to their ability to convert chemical energy directly into electricity at up to 60% efficiency, operate with a wide range of fuels from hydrogen to biogas, and emit virtually no pollutants. SOFC usually operates under high temperature conditions and can directly convert the chemical energy in the fuel into electrical energy with high efficiency. Its core structure is composed of ceramic electrolytes, electrode materials, etc. In recent years, with the advancement of material technology and the improvement of system integration capabilities, SOFC has begun to be used in many scenarios such as household energy systems, industrial power generation and aerospace equipment. However, the technology still faces challenges such as high cost, poor thermal stability and slow startup. This article analyzes the basic structure and operation mechanism of SOFC, introduces the development trend of its key materials, evaluates current typical applications, and explores possible technological breakthroughs in the future, providing ideas and references for promoting sustainable energy systems.

Keywords: solid oxide fuel cell; clean energy; structural design; material development;

1. Introduction

Solid oxide fuel cell (SOFC) is a high-temperature power generation device, and the electrolyte is made of ceramic materials that can conduct oxygen ions [1]. Its working principle does not rely on traditional combustion, but uses the electrochemical reaction between oxygen and fuel to directly generate electricity [2]. At the anode, the fuel is oxidized; at the cathode,

the oxygen is reduced to oxygen ions. The oxygen ions pass through the electrolyte, react with the fuel to generate water or carbon dioxide, and release electrons to output current.

In recent years, as the global demand for low-carbon energy continues to increase [3], SOFC technology has gradually attracted more attention [4]. It not only has high conversion efficiency, but also has low requirements for fuel types, and has almost no pol-

lutant emissions. It is suitable for use in combination with various energy sources such as natural gas and biogas. At present, SOFC has shown certain application potential in the fields of distributed power generation and industrial waste heat recovery [5], and is also considered to be one of the important directions for promoting the development of clean energy in the future.

Therefore, this paper aims to provide a comprehensive overview of SOFC technology by introducing its working principles and structural types, evaluating its current technical status and real-world applications, and discussing recent innovations and future development trends.

2. Overview of SOFC

2.1 Working principle of SOFC

Solid oxide fuel cells (SOFC) rely on high-temperature migration of oxygen ions to achieve electrical energy output during operation. Usually at 600 to 1000 degrees Celsius [6], the cathode absorbs oxygen and converts it into oxygen ions (O^{2-}). These ions pass through the ceramic electrolyte membrane and react with fuel (such as hydrogen, natural gas or carbon monoxide) at the anode to produce water or carbon dioxide, while releasing electrons. These electrons flow through the external circuit to form a current and achieve electrical energy output [7].

Unlike traditional thermal engine systems that rely on combustion, SOFC uses a direct electrochemical conversion process without a combustion step in the middle [1], so the energy conversion efficiency is higher. The high temperature characteristics also bring additional advantages, such as the ability to output heat energy at the same time, achieve combined heat and power, or use the reaction heat for internal reforming of the fuel, thereby improving the overall efficiency of the system.

2.2 Types of SOFC

Based on different structural designs and application scenarios, SOFC can be divided into the following common forms:

2.2.1 Tubular SOFC

This type uses a hollow tubular structure to coat the anode, electrolyte and cathode in sequence on the outer surface of a ceramic support tube [8]. It has good stability and strong sealing, and is suitable for medium and large-scale continuously operated power generation systems. However, due to the complex manufacturing process, the power density per unit volume is low and the cost is relatively high.

2.2.2 Planar SOFC

This type of fuel cell is pressed from multiple layers of materials, with the anode, electrolyte and cathode layers arranged in parallel and stacked to form a battery module. The flat plate has a compact structure and high-power density [9]. It is one of the most mature forms of research and commercialization at present, and is widely used in household thermoelectric devices and distributed energy systems.

2.2.3 Intermediate-Temperature SOFC

Traditional SOFC has a high operating temperature, strict material requirements and a long hot start time [10]. To improve stability and reduce system pressure, researchers have developed medium-temperature SOFCs that operate at 500–700°C [10]. By using new electrolytes such as doped cerium oxide (GDC) or scandium stabilized zirconia (LSGM) and optimizing the electrode structure, it is possible to effectively extend the service life while taking into account performance, and also provide possibilities for wider commercial applications.

3. Current Status of SOFC Technology

3.1 Life and Stability

When SOFC is running for a long time, the biggest risk comes from the impact of high temperature environment on the internal materials of the battery. The higher the temperature, the greater the structural stress between the electrolyte, anode and cathode, and the material is prone to delamination, deformation, and even microcracks [11]. Moreover, chemical corrosion is more serious when the atmosphere changes alternately between air, hydrogen, methane, etc.

In order to deal with these problems, researchers have developed many new materials. For example, the team of the University of Tokyo in Japan studied yttrium-doped zirconium oxide (YSZ) and lanthanum-doped cerium oxide (GDC) mixed electrolytes and found that they can still maintain high ionic conductivity at 700°C [12], and the thermal stability is much better than traditional materials.

In practical applications, the commercial SOFC battery stack launched by Bloom Energy in the United States has a target life of 5-10 years and a cumulative operating time of more than 40000 hours [13]. The Fraunhofer Institute in Germany has also successfully achieved continuous operation of SOFC in microgrid systems for more than 3 years in multiple demonstration projects [14]. Although this does not mean that the conditions for full popularization are ripe, it at least shows that SOFC has made significant breakthroughs in long-term stability.

3.2 Start-up time

Compared to low-temperature fuel cells (such as PEMFC or AFC), one of SOFC's biggest disadvantages is its slow startup. This is primarily because its reaction process relies on the migration of oxygen ions in the electrolyte, a process that must be above 600°C to be effective [15]. Traditional ceramic heating methods typically take several hours to heat from room temperature to operating temperature, making them unsuitable for scenarios requiring rapid response, such as emergency power supplies or on-board systems. To address this issue, there are currently two main technical approaches: one is to develop medium-temperature SOFC systems with an operating temperature of 500–700°C, and the other is to introduce rapid heating devices into the system, such as infrared preheaters or embedded heating resistors [16].

For example, the SteelCell® technology of Ceres Power in the UK uses a metal support structure and GDC electrolyte to compress the start-up time to 30–45 minutes [17], which is suitable for home and small and medium-sized scenarios. The Ene-Farm home system developed by Osaka Gas and Kyocera in Japan can also complete hot start in about 1 hour [18]. More than 400,000 sets have been installed in Japan, becoming one of the world's largest SOFC commercial cases [19].

3.3 Safety and control system

Although SOFCs operate without open flames, they still require significant safety considerations due to the extremely high temperatures and the presence of flammable gases such as hydrogen and methane. In particular, if the system is not properly sealed, air and fuel may mix. Localized overheating could lead to thermal diffusion, potentially leading to leakage or explosion, with significant consequences.

Current mainstream SOFC products are generally equipped with a variety of sensors and control systems to ensure stable and safe operation [20]. For example, temperature probes monitor the stack for hot spots in real time; pressure and flow sensors prevent imbalances in the fuel supply; and software automatically adjusts the load and gas concentration, even implementing protective shutdowns when necessary. Furthermore, a BMS module similar to that used in electric vehicles monitors key parameters such as operating status, voltage, and current.

For example, Bloom Energy's Energy Server features multiple layers of redundant protection, including automatic stack shutdown, automatic gas shutoff, and cloud-based remote monitoring [21]. Ceres Power's equipment focuses on "plug and play" and remote diagnosis. Users can even view the real-time status of the equipment di-

rectly through the mobile app, making operation more convenient and safer.

3.4 Efficiency and application performance

The biggest highlight of SOFC is its high energy conversion efficiency. In the single power generation mode, its electrical efficiency can reach 50–60% [22], which is much higher than that of traditional internal combustion engines (about 25–35%). If the system also recovers heat energy for heating, hot water and other heat loads, the total efficiency can even exceed 80–90% [23], which makes SOFC particularly suitable for household combined heat and power (CHP) systems.

For example, Bloom Energy's server systems can achieve an electrical efficiency of approximately 60% in power generation mode; if excess heat is used for heating or hot water, the overall efficiency can be increased to 85% [24]. Ceres Power, in a collaborative project with China National Petroleum Corporation, has also launched a micro-energy station solution for homes [25]. This integrates SOFCs and thermal storage tanks, enabling not only power generation but also heating and domestic hot water needs, achieving the goal of "small system, high energy efficiency."

SOFC can also flexibly use a variety of fuels, such as hydrogen, natural gas, methanol and even biogas, which is very suitable for use with existing natural gas pipelines, and is also expected to provide stable and reliable clean energy for rural and remote areas.

4. Technological Innovation and Future Development of SOFC

4.1 Breakthroughs in new materials and structural design

Most traditional SOFC electrolytes use YSZ (yttrium-stabilized zirconia). Although it has good stability, it must be above 800°C to show sufficient oxygen ion conductivity, which brings about problems such as difficult thermal management, slow startup, and high structural stress [26]. To solve this problem, many researchers have begun to explore new electrolyte materials required for medium-temperature SOFC (500–700°C).

For example, the National Energy Laboratory (NETL) and MIT and other institutions have studied a class of materials called LSGM (lanthanum-doped scandium magnesium oxide), which has a conductivity at 600°C that can reach the level of YSZ at 900°C [27]. GDC (lanthanum-doped cerium oxide) is also widely used in medium-temperature systems because of its high oxygen ion mobility. Howev-

er, GDC is prone to $\text{Ce}^{4+} \rightarrow \text{Ce}^{3+}$ reaction in a reducing atmosphere, resulting in increased electronic conductivity. Currently, it is mostly solved by combining with other materials [28].

In terms of electrode materials, perovskite cathode materials (such as LSCF and BSCF) show good oxygen reduction reaction activity and are gradually replacing the early LSM. The anode is developing from the traditional Ni-YSZ to Ni-GDC and composite porous anodes [29], which helps to improve gas diffusion and resistance to carbon deposition.

In terms of structure, Ceres Power's SteelCell® technology uses metal supports to replace traditional ceramic supports [30], which is not only lightweight, but also greatly reduces the risk of damage caused by thermal expansion and contraction. This design makes SOFC more suitable for batch manufacturing and modular integration.

4.2 Upgrade of intelligent control and system integration

Modern SOFC systems not only rely on the materials themselves, but also increasingly rely on the optimization of the “software layer”. Artificial intelligence (AI) and machine learning have been introduced into SOFC modules such as temperature control, fuel management, and stack diagnosis, which can automatically adjust parameters according to the real-time operating status to improve system stability and response speed [31].

For example, Fraunhofer IKTS in Germany has developed an AI-based thermal field learning model that can predict the temperature distribution of different areas inside the stack and identify the risk of “hot spots” in advance [32]. Ceres Power has also deployed a remote monitoring platform in its system, and users can view key indicators such as battery status, efficiency, and gas consumption through a mobile phone app [33].

In terms of system integration, SOFC is also increasingly inclined to “multifunctionality”. It can not only generate electricity, but also provide heating and hot water, and even participate in microgrid regulation. The integrated energy server (Bloom Box) launched by Bloom Energy has been used in data centers or office parks of companies such as Google, Apple, and Walmart, achieving clean, stable, and efficient energy supply [34].

4.3 Commercialization progress and future development trends

In the past, SOFC was considered to be only a cutting-edge technology in the laboratory, but now it has begun to enter the market. Users only need to connect to city gas to generate electricity, provide hot water and heating

at the same time, which is energy-saving and convenient. Bloom Energy has become one of the most commercially successful companies in Europe and the United States, with its products deployed in over 700 enterprises [35]. While its system is relatively expensive, it offers low operational costs, a long lifespan, and ease of maintenance, making it suitable for locations requiring high stability. Looking ahead, the future development of SOFC technology is likely to focus on several key directions. First, its compatibility with both hydrogen and natural gas positions it well for a hydrogen-based energy economy, where clean and efficient power generation will be in high demand. Moreover, the stable and continuous electricity output of SOFCs makes them an ideal partner for intermittent renewable sources such as wind and solar, helping to smooth fluctuations and improve overall grid reliability. Another promising feature is the scalability of SOFC systems — from compact residential units to large-scale industrial installations, their modular design allows for flexible deployment across various energy needs. Lastly, while high material costs currently remain a major barrier, ongoing advancements in materials science and increased manufacturing volumes are expected to drive down prices, paving the way for broader commercial adoption.

5. Conclusion

In summary, solid oxide fuel cells (SOFCs) are no longer confined to laboratories. With their high efficiency, fuel flexibility, and nearly zero emissions, they are becoming a practical and promising solution for clean energy applications. While technical barriers such as high temperature operation, slow startup, and material costs still exist, ongoing advances in materials science, structural engineering, and intelligent system control are steadily pushing the technology forward. Commercial deployments by companies such as Bloom Energy, Ceres Power, and Kyocera demonstrate that SOFCs can meet real-world energy demands — whether in residential, industrial, or backup power systems. Their compatibility with both hydrogen and conventional fuels also makes them a strategic fit for future integrated energy networks. As the world accelerates toward carbon neutrality, SOFCs are well-positioned to play a key role in building a more efficient, reliable, and sustainable energy future.

References

- [1] Minh, N. Q. Solid oxide fuel cell technology—features and applications. *Solid State Ionics*, 2004, 174(1), 271–277.
- [2] STEELE, B. C. H., & HEINZEL, A. Materials for fuel-cell technologies: Materials for clean energy. *Nature*, 2001,

414(6861), 345–352.

[3] Liang, Y., Kleijn, R., & Van der Voet, E. Increase in demand for critical materials under IEA Net-Zero emission by 2050 scenario. *Applied Energy*, 2023, 346, 121400.

[4] Shao, Z., & Haile, S. M. A high-performance cathode for the next generation of solid-oxide fuel cells. *Nature*, 2004, 431(7005), 170–173.

[5] Talukdar, A., Chakrovorty, A., Sarmah, P., Paramasivam, P., Kumar, V., Yadav, S. K., Manickam, S., & Ahmed, M. A Review on Solid Oxide Fuel Cell Technology: An Efficient Energy Conversion System. *International Journal of Energy Research*, 2024, 6443247, 1-20.

[6] Huang, K., & Goodenough, J. B. Solid oxide fuel cell technology: principles, performance and operations, 2009.

[7] Zakaria, Z., Awang Mat, Z., Abu Hassan, S. H., & Boon Kar, Y. A review of solid oxide fuel cell component fabrication methods toward lowering temperature. *International Journal of Energy Research*, 2020, 44(2), 594-611.

[8] Huang, K., & Singhal, S. C. Cathode-supported tubular solid oxide fuel cell technology: A critical review. *Journal of power sources*, 2013, 237, 84-97.

[9] Herring, J. S., O'Brien, J. E., Stoots, C. M., Hawkes, G. L., Hartvigsen, J. J., & Shahnam, M. Progress in high-temperature electrolysis for hydrogen production using planar SOFC technology. *International Journal of Hydrogen Energy*, 2007, 32(4), 440-450.

[10] Brett, D. J., Atkinson, A., Brandon, N. P., & Skinner, S. J. Intermediate temperature solid oxide fuel cells. *Chemical Society Reviews*, 2008, 37(8), 1568-1578.

[11] Shin, J. S., Saqib, M., Jo, M., Park, K., Park, K. M., Ahn, J. S., ... & Park, J. Y. Degradation mechanisms of solid oxide fuel cells under various thermal cycling conditions. *ACS applied materials & interfaces*, 2021, 13(42), 49868-49878.

[12] Chou, J. T., Inoue, Y., Kawabata, T., Matsuda, J., Taniguchi, S., & Sasaki, K. Mechanism of SrZrO₃ formation at GDC/YSZ interface of SOFC cathode. *Journal of The Electrochemical Society*, 2018, 165(11), F959.

[13] Bulushi, F. A., AM, S., Patil, G., & Walke, S. Studies on bloom energy server. *Recent Innovations in Chemical Engineering (Formerly Recent Patents on Chemical Engineering)*, 2022, 15(3), 214-225.

[14] Fraunhofer IKTS. Degradation study on SOC stacks. Fraunhofer Institute for Ceramic Technologies and Systems IKTS, 2024.

[15] Jeong, H., Lee, C., Son, J. W., Lee, S. Y., Yoon, K. J., Shin, D., ... & Kim, H. Advancing towards ready-to-use solid oxide fuel cells: 5 minute cold start-up with high-power performance. *Journal of Materials Chemistry A*, 2023, 11(14), 7415-7421.

[16] Li, J., Cheng, J., Zhang, Y., Chen, Z., Nasr, M., Farghali, M., ... & Osman, A. I. Advancements in solid oxide fuel cell technology: bridging performance gaps for enhanced

environmental sustainability. *Advanced Energy and Sustainability Research*, 2024, 5(11), 2400132.

[17] Xu, H., Han, Y., Zhu, J., Ni, M., & Yao, Z. Status and progress of metal-supported solid oxide fuel cell: Towards large-scale manufactory and practical applications. *Energy Reviews*, 2024, 3(1), 100051.

[18] WEIDNER, R. E., ORTIZ, C. R., & DAVIES, J. Global deployment of large capacity stationary fuel cells, 2019.

[19] Simader, G., & Vidovic, P. Success factors for demonstration projects of small-scale stationary fuel cells in residential buildings. In *E3S Web of Conferences*, 2022, 334, 04007.

[20] Li, M., Chen, Z., Dong, J., Xiong, K., Chen, C., Rao, M., ... & Peng, J. A data-driven fault diagnosis method for solid oxide fuel cell systems. *Energies*, 2022, 15(7), 2556.

[21] Elkafas, A. G., Rivarolo, M., Gadducci, E., Magistri, L., & Massardo, A. F. Fuel cell systems for maritime: a review of research development, commercial products, applications, and perspectives. *Processes*, 2022, 11(1), 97.

[22] Corigliano, O., Pagnotta, L., & Fragiaco, P. On the technology of solid oxide fuel cell (SOFC) energy systems for stationary power generation: A review. *Sustainability*, 2022, 14(22), 15276.

[23] Sadhukhan, J., et al. Thermodynamic models of SOFCbased microCHP systems achieving 85% overall efficiency. *Sustainability*, 2023, 16(23), 10773.

[24] Bloom Energy. Bloom Energy Announces Hydrogen Solid Oxide Fuel Cell with 60% Electrical Efficiency and 90% HighTemperature Combined Heat and Power Efficiency. *Business Wire*, 2024.

[25] Li, J., Cheng, J., Zhang, Y., Chen, Z., Nasr, M., Farghali, M., ... & Osman, A. I. Advancements in solid oxide fuel cell technology: bridging performance gaps for enhanced environmental sustainability. *Advanced Energy and Sustainability Research*, 2024, 5(11), 2400132.

[26] Solovyev, A., Shipilova, A., & Smolyanskiy, E. Solid oxide fuel cells with magnetron sputtered single-layer SDC and multilayer SDC/YSZ/SDC electrolytes. *Membranes*, 2023, 13(6), 585.

[27] Xie, M., Cai, C., Duan, X., Xue, K., Yang, H., & An, S. Review on Fe-based double perovskite cathode materials for solid oxide fuel cells. *Energy Materials*, 2024, 4, 400007.

[28] Brett, D. J., Atkinson, A., Brandon, N. P., & Skinner, S. J. Intermediate temperature solid oxide fuel cells. *Chemical Society Reviews*, 2008, 37(8), 1568-1578.

[29] Ranjbar, R., Yadollahi Farsani, F., & Safarpour Dehkordi, F. Antimicrobial resistance and genotyping of *vacA*, *cagA*, and *iceA* alleles of the *Helicobacter pylori* strains isolated from traditional dairy products. *Journal of Food Safety*, 2019, 39(2), e12594.

[30] Zhang, Z., Du, H., Xu, K., Zhang, X., Ma, X., & Shuai, S. Review of the application of metal-supported solid oxide fuel

cell in the transportation field. *Automotive Innovation*, 2025, 8, 443–471.

[31] Corigliano, O., Pagnotta, L., & Fragiaco, P. On the technology of solid oxide fuel cell (SOFC) energy systems for stationary power generation: A review. *Sustainability*, 2022, 14(22), 15276.

[32] Fraunhofer IKTS. Modeling and simulation of SOFC systems using AI-based diagnostics, 2024.

[33] Ceres Power. The AI energy challenge: SOFC solutions for data centres, 2025.

[34] Bloom Energy. How distributed generation is helping data centers overcome energy challenges in the AI era. *Data Center Frontier* sponsored article, 2025.

[35] Bloom Energy. Onsite generation expected to fully power 27% of data center facilities by 2030. *MidYear 2025 Data Center Power Report*. Retrieved from Bloom Energy website, 2025.